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Manuscript Number: JDDST-D-22-01553

Enhanced localization of cefazoline sodium in the ocular tissue using thermosensitive-mucoadhesive hydrogels: Formulation development, hemocompatibility and in vivo irritation studies

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Enhanced localization of cefazoline sodium in the ocular tissue using thermosensitive-mucoadhesive hydrogels: Formulation development, hemocompatibility and in vivo irritation studies

Dear Dr. Permana,

Thank you for submitting your manuscript to Journal of Drug Delivery Science and Technology.

I have completed my evaluation of your manuscript. The reviewers recommend reconsideration of your manuscript following major revision. I invite you to resubmit your manuscript after addressing the comments below. Please resubmit your revised manuscript by Sep 29, 2022.

When revising your manuscript, please consider all issues mentioned in the reviewers' comments carefully: please outline every change made in response to their comments and provide suitable rebuttals for any comments not addressed. Please note that your revised submission may need to be re-reviewed.

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Shirui Mao
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Journal of Drug Delivery Science and Technology

Editor and Reviewer comments:

Reviewer #1: An interesting piece of work that adds new knowledge to the field and will, in due course, be widely read and cited. All experiments have been well-planned and interpreted, data analysis is appropriate and conclusions drawn sensible.

The authors should consider the RaScaL study (Suner et al Ophthalmologica. 2014 Nov 26. doi: 10.1159/000367902.) Would the approaches there provide any insights into the current work on ocular drug delivery?

Could the authors possibly provide details of how this work will be translated? How would the system be manufactured at scale? How would patients/clinicians use it? What barriers to success exist? How would it be sterilised without damaging the cargo or crosslinking the polymers? Such assertions are important for readers to understand the potential impact of the work.

The Figure legends are not labelled with means \pm S.D. What is n ? Such attention to detail is important. Similarly, many words derived from classical languages (e.g. *in situ*) should be in italics throughout.

Finally, the authors should re-read and improve the scientific English throughout.

Reviewer #2: The drug delivered by the mucoadhesive-thermosensitive gel was an antibiotic, a microbiological assay will be needed to demonstrate the activity of the drug is maintained.

In the ex vivo permeation studies presented in the section 3.7 needs to be analysed in terms of permeation and not using the release kinetic as presented in table 5. To determine the permeation rates and the flux please see the following papers (<https://doi.org/10.3390/md15120370>; <https://doi.org/10.1016/j.ijpharm.2020.119020>) to determine those parameters and discuss data considering those parameters.

There are many data repeated in graphics and, in the text, please delete some of the graphics namely the Figure 1, Figure 2c; Figure 4a.

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Enhanced localization of cefazoline sodium in the ocular tissue using thermosensitive-mucoadhesive hydrogels: Formulation development, hemocompatibility and in vivo irritation studies

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We have received the above referenced manuscript you submitted to Journal of Drug Delivery Science and Technology.

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Manuscript Number: JDDST-D-22-01553

Enhanced localization of cefazolin sodium in the ocular tissue using thermosensitive-mucoadhesive hydrogels: Formulation development, hemocompatibility and in vivo irritation studies

Response to Reviewers

We are very thankful to the expert reviewers for taking the time to kindly review our manuscript and provide helpful comments for improvement and clarification. We have made some changes to the manuscript as a result of these comments. We believe that the manuscript is now substantially improved. We have addressed each of the reviewers' comments in detail below. Importantly, we have made a great effort to improve the English and the discussion parts of our revised manuscript.

Reviewer #1: An interesting piece of work that adds new knowledge to the field and will, in due course, be widely read and cited. All experiments have been well-planned and interpreted, data analysis is appropriate and conclusions drawn sensible.

Response to Reviewer

We are very thankful to the Reviewer for taking the time to review this manuscript and for the expert review, providing helpful comments. We are glad that the Reviewer thinks that our work is interesting. We have made a number of key changes to the manuscript as a result of these comments. We believe that the manuscript is now substantially improved. We have addressed each one of the reviewers' comments in detail.

The authors should consider the RaScaL study (Suner et al Ophthalmologica. 2014 Nov 26. doi: 10.1159/000367902.) Would the approaches there provide any insights into the current work on ocular drug delivery?

Response:

We thank the reviewer for the suggestion. The paper evaluated the efficacy, durability, and safety of a single treatment with intravitreal ranibizumab plus peripheral scatter laser (RaScaL) in patients with diabetic macular edema associated with peripheral retinal nonperfusion on ultrawide-field fluorescein angiography (UWFA). Although the paper did not describe the ocular drug delivery, the paper provides several useful information with regard to the design of clinical study in phase I/II. The clinical study design described could be potentially applied when the system developed in our study reaches the clinical evaluations.

Could the authors possibly provide details of how this work will be translated? How would the system be manufactured at scale? How would patients/clinicians use it? What barriers to success exist? How would it be sterilised without damaging the cargo or crosslinking the polymers? Such assertions are important for readers to understand the potential impact of the work.

Response:

We thank the reviewer for the suggestion. Indeed, we agree that the explanations recommended by the Reviewer is crucial. As a result of the reviewer suggestion, we have added the following explanation in the revised manuscript:

The use thermosensitive hydrogel in ocular drug delivery have been widely explored. However, several aspects should be considered. As this was intended to ocular administration, before moving to the industrial steps, sterilization process should be developed. Several studies have shown that Pluronic[®]-based hydrogels could be sterilized using steam heat sterilization with autoclave (Rafael et al., 2019; Raval and Bagada, 2021). However, due to the instability of CFZ in high temperature, gamma radiation could be the appropriate method for the sterilization. The development of this system is still in the early step. Therefore, it is also crucial to further optimize the formulation to ensure that the hydrogel possess adequate properties after the administration to the patients. As the formulation is in the liquid form, the hydrogel could be administered using the same method as the eye drops. Additionally, the information regarding the transformation of the liquid to the gel after the administration should be given to the patients. However, prior

to the application in the clinic, further study investigating *in vivo* pharmacokinetic and pharmacodynamic studies in appropriate animal models should now be conducted.

The Figure legends are not labelled with means +/- S.D. What is n= ? Such attention to detail is important. Similarly, many words derived from classical languages (e.g. *in situ*) should be in italics throughout.

Response:

We thank the reviewer for pointing this out. We have added this information in the revised manuscript. Additionally, all words from classical languages are now in italics.

Finally, the authors should re-read and improve the scientific English throughout.

Response:

We thank the reviewer for the suggestion. We have re-read the manuscript and have made a great effort to improve the English throughout.

Reviewer #2: The drug delivered by the mucoadhesive-thermosensitive gel was an antibiotic, a microbiological assay will be needed to demonstrate the activity of the drug is maintained.

Response:

We are very thankful to the expert reviewers for taking the time to kindly review our manuscript and provide helpful comments for improvement and clarification. As a result of this comment, we have performed *in vitro* antimicrobial activity against *Pseudomonas aeruginosa*. It was found that the inhibition zones of CFZ solution and CFZ in thermosensitive *in situ* gel were 21.98 ± 1.98 mm and 20.76 ± 1.21 mm. Analyzed statistically, there was no a significant different ($p > 0.05$) between those values, indicating that the formulation did not affect the antimicrobial activity of CFZ against *Pseudomonas aeruginosa*. We have added this in the revised manuscript.

In the ex vivo permeation studies presented in the section 3.7 needs to be analysed in terms of permeation and not using the release kinetic as presented in table 5. To determine the permeation rates and the flux please see the following papers (<https://doi.org/10.3390/md15120370>; <https://doi.org/10.1016/j.ijpharm.2020.119020>) to determine those parameters and discuss data considering those parameters.

Response:

We thank the reviewer for the suggestions. We have calculated permeation flux and permeation coefficient from our study according to the suggested papers (Silva et al., 2020, 2017) and have included the results in the revised manuscript.

There are many data repeated in graphics and, in the text, please delete some of the graphics namely the Figure 1, Figure 2c; Figure 4a.

Response:

We thank the reviewer for the suggestion. We have deleted some graphs as requested.

References

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<https://doi.org/10.3390/md15120370>

1 **Enhanced localization of cefazolin sodium in the ocular tissue using thermosensitive-**
2 **mucoadhesive hydrogels: Formulation development, hemocompatibility and *in vivo***
3 **irritation studies**

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12

13

ABSTRACT

14 Bacterial keratitis is an eye infectious disease which became a global concern due to its
15 impact to visual impairment in worldwide. Cefazolin (CFZ) is the first-line drug for the
16 treatment of bacterial keratitis and only available in the drop dosage form. However, the
17 administration of eye drops results in lack of bioavailability, which is below 5%. Therefore,
18 an innovation is required in an attempt to overcome these problems. In this study, the
19 development of a mucoadhesive-thermosensitive gel *in situ* preparation was carried out. A
20 combination of Pluronic F127 and F68 was used as a thermosensitive agent. To increase the
21 contact time, a mucoadhesive agent, namely hyaluronic acid was added to the formulation.
22 Several steps of evaluation were performed to ensure that the developed formula possessed
23 desired characteristics, including determination of gelation temperature, pH test, viscosity
24 test, rheology test, mucoadhesive test, drug content test, *in vitro* drug release test, *in vivo*
25 irritation test on experimental animals, *ex vivo* permeation test, and hemolysis test. The
26 results that the formulations developed exhibited desired characteristics, with gelation
27 temperatures of around 37°C. The formulation could also control the release and improve the
28 localization of CFZ in the ocular tissue compared to control solution. Furthermore, the
29 incorporation of CFZ into this approach did not change the antimicrobial activity of CZ
30 against *Pseudomonas aeruginosa*. Importantly, no toxicity and irritation were observed after
31 the application of this approach. However, further research is needed to evaluate the
32 pharmacokinetic and pharmacodynamic in the appropriate animal models.

33 **Keywords:** Cefazoline, Bacterial keratitis, mucoadhesive-thermosensitive *in situ* gel

34 1. Introduction

35 Bacterial keratitis is an inflammatory process caused by a bacterial infection.
36 Approximately 64.6% of keratitis cases are caused by bacterial infection. The use of contact
37 lenses has been reported to be the main risk factor of this disease [1]. Previous study has
38 reported that the incidence of microbial keratitis with stromal infiltration and ulcers due to
39 the use of contact lenses was 38,7% [2]. Contact lenses block the cornea from oxygen, tears,
40 and ocular secretions; which can affect the occurrence of infection or inflammation of the
41 corneal layer, resulting in an impaired vision. The symptoms and signs of bacterial keratitis
42 are pain, redness, blurred vision, discharge, corneal infiltrates, ulceration, photophobia, and
43 inflammation of the anterior chamber of the eye. The treatment of this disease must be
44 supported by a medical history, physical examination and slit-lamp examination [3].

45 Topical antibiotics remain the first-line treatment for bacterial keratitis [4]. Cefazoline
46 (CFZ) is a first-generation cephalosporin antibacterial used for the treatment of infections,
47 inhibiting bacterial cell wall synthesis. CFZ is given in the form of eye drops at a dose of 50
48 mg/mL [3]. In ocular drug application, the presence of precorneal factors and anatomic
49 barriers have limited the bioavailability of the eye preparations. Precorneal factors include
50 solution drainage, loss from blinking, tear film formation, and increased tear secretion.
51 Considering all these precorneal factors, the contact time of the ocularly applied preparations
52 is very low and, therefore, only <5% can penetrate the intraocular tissues [5]. Eye drops are
53 the type of preparation that are most often applied topically to the ocular surface to treat
54 external ocular infections. However, due to the protective mechanism of the eye, it results in
55 very poor bioavailability of the drug [6]. Furthermore, it also causes low drug permeation to
56 the ocular tissue and because the eye pocket has a limited capacity, the amount of drug
57 absorbed by the ocular tissue is unknown [7].

58 The *in situ* gel delivery system is concerned with the conversion of the liquid state of
59 the formulation (solution) into gel at the site of application under certain physiological
60 conditions. Many factors regulate gel formation *in situ* including temperature, pH, solvent
61 exchange, ionic cross-linking and ultraviolet radiation [8]. Thermosensitive *in situ* gel is a

62 vicious liquid that can thermally turn into a gel after entering the ocular physiological fluid
63 [9]. The heat-sensitive *in situ* gelling system is in a liquid state at room temperature (25°C)
64 and undergoes a transition to a gel at a temperature value close to physiological body (32°C
65 - 37°C), depending on the site of administration [10]. The main advantages of *in situ* gel are
66 delayed and controlled drug delivery, and reduced or absent blurred vision **in comparison with**
67 **eye** ointments. Another advantage of the *in situ* gel system over eye drops and ointments is
68 the increased bioavailability of the drug due to increased pre-corneal contact. In addition, the
69 *in situ* gel system may be more convenient than insoluble or soluble insertions [11].

70 The polymer commonly used in thermosensitive *in situ* gel systems is Poloxamer
71 (Pluronic®) [12]. Pluronic F127 is the type of Poloxamer that has been most often used in
72 ophthalmic delivery systems [13]. It was found that *in situ* gel formulation using single
73 Poloxamer F127 had $27.2 \pm 0.4^\circ\text{C}$ of gelation temperature. This temperature was lower than
74 the physiological temperature of the eye. The gelation temperature of Pluronic® F127 can be
75 increased by combining Pluronic F127 with other types of Pluronic [14]. To the best of our
76 knowledge, until now, there has been no research that examines Pluronic® F68 as a single
77 polymer or in combination with other Pluronic for *in situ* gel preparations for the eye
78 containing CFZ.

79 To enhance the contact time of the *in situ* gel with the ocular tissue for potential
80 increased bioavailability, a mucoadhesive agent is needed in the preparation [15]. One of the
81 mucoadhesive agents that most recommended is hyaluronic acid (HA). HA has the advantage
82 when formulated into ophthalmic preparations, mainly due to its ability to increase contact
83 time. Importantly, HA has excellent biocompatibility and biodegradability in the eye. This
84 polymer can even protect the corneal epithelium from dehydration and reduce the
85 inflammatory response to dehydration. Also, HA can lubricate the ocular surface to prevent
86 dehydration [16]. In this study, a thermosensitive mucoadhesive *in-situ* gel formulation from
87 CFZ was carried out using a combination of Pluronic® F127 and F68 with HA as
88 mucoadhesive agent.

89

90

91 **2. Material and Methods**

92 **2.1. Materials**

93 Cefazolin sodium salt (CFZ) (purity 89.1–110.1%) was purchased from Sigma Aldrich Pte
94 Ltd. (Singapore, Singapore). Pluronic® F127 and F68 used in this study were generous gifted
95 by BASF Indonesia (Jakarta, Indonesia). Other chemicals used in this study were analytical
96 grade

97 **2.2. Design of Formulation**

98 The formulation of the thermosensitive *in situ* gel was produced by dissolving Pluronic®
99 F127 and Pluronic® F68 at a temperature of 4°C until fully dissolved [17]. The mucoadhesive
100 agent was prepared by carefully weighing the HA, then dissolved in distilled water and
101 carefully heated on an electric stove until dissolved. After the hyaluronic was completely
102 dissolved and placed at room temperature, it was mixed with the thermosensitive gel
103 formulation. CFZ and benzalkonium chloride were weighed carefully and added to the basic
104 mixture of the mucoadhesive thermosensitive *in situ* gel formulation, then homogenized.
105 Furthermore, several evaluations of the mucoadhesive thermosensitive *in situ* gel formula
106 were carried out.

107 **Table 1.** Design of CFZ mucoadhesive-thermosensitive formulation

Ingredients %(w/w)	F1	F2	F3	F4	F5
CFZ Sodium	0,35	0,35	0,35	0,35	0,35
Pluronic® F127	15	15	15	15	15
Pluronic® F68	5	5	5	5	5
HA	0,25	0,2	0,15	0,1	0
Benzalkonium Chloride	0,01	0,01	0,01	0,01	0,01
Distilled water	Ad 100	Ad 100	Ad 100	Ad 100	Ad 100

108 **2.3. Evaluation**

109 **2.3.1. Mucoadhesive-thermosensitive *in situ* gelation temperature study**

110 Initially, 2 mL of each formula was put into test tube. The test tube was put into water
111 bath at 20°C and then the temperature of the bath was increased slowly until 65°C. The test
112 tubes containing each formula were observed visually after a 1°C increase in temperature.

113 The gelation temperature was recorded as the temperature at which the gel did not move
114 when the test tube was inverted to 90° [18].

115 **2.3.2. Mucoadhesive study of CFZ *in situ* gel formulation**

116 Determination of the mucoadhesive test was carried out using a modified balance tool,
117 where the left arm of the balance was attached to two layers of pig eye tissue that were applied
118 with mucoadhesive thermosensitive *in situ* gel formula. On the other hand, on the right arm
119 of the balance tool, 1 gram of the metal weight was added every 30 seconds until ocular tissue
120 was separated from other tissue [19,20]

121 **2.3.2. pH Evaluation**

122 The pH value was assessed using a digital pH meter.

123 **2.3.3. Viscosity rheology study of gel *in situ* formulation**

124 Rheological was determined by using viscometer (Brookfield, USA). In an attempt to
125 assess the flow properties, we calculated the velocity against the viscosity. The viscosity was
126 measured on three different temperatures, namely 25°C, 37°C, and 4°C.

127 **2.4. Drug content determination of CFZ in mucoadhesive-thermosensitive *in situ* gel** 128 **formulation**

129 Each formula was carefully weighed and dissolved in simulated tear fluid (STF) to
130 achieve a final concentration of 10 µg/mL. All formulations were then centrifuged for 15
131 minutes and the supernatant was measured using UV-vis spectrophotometry at a wavelength
132 of 233.4 nm [21].

133 **2.5. *In vitro* drug release study**

134 In this study, the release study of CFZ from mucoadhesive-thermosensitive *in situ* gel
135 formulation, a dialysis method was utilized. The experiment was carried out at 37°C at 100
136 rpm. The dialysis membrane used was soaked with STF prior to the experiments. Afterwards,
137 the amount of each formula was calculated to be equivalent to 10 mg of CFZ, then put into a
138 dialysis membrane and both sides of the membrane were closed with airtight clamps. The
139 membrane was put into 100 mL of STF inside the glass bottle. The media (1 mL) was then
140 sampled at a time range of 0.25, 0.5, 0.75, 1, 2, 3, 4, 5, 6, 7, 8, and 24 hours and replaced

141 with 1 ml of fresh STF. The release kinetic profile was determined using the following
142 equations [22].

143 *Zero Order Kinetics: $C_t = C_0 + k_0t$*

144 *First Order Kinetics: $\ln C_t = \ln C_0 + k_1t$*

145 *Higuchi Model: $C_t = k_H t$*

146 *Korsmeyer–Peppas Model: $C_t = k_{KP} t^n$*

147 *Hixson–Crowell Model: $C_t^{1/3} = C_0^{1/3} + k_{HC} t$*

148 Where C_t is the amount of CFZ released at time t , C_0 is the initial concentration of CFZ
149 in STF (at $t=0$), k_0 is constant of zero-order kinetics, k_1 is constant of first-order kinetics, k_H
150 is constant of Higuchi model, k_{KP} is constant of Korsmeyer-Peppas model, and k_{HC} is
151 constant of Hixson-Crowell model.

152 **2.6. *Ex vivo* permeation study of CFZ mucoadhesive-thermosensitive *in situ* gel** 153 **formulation**

154 Pig cornea was taken carefully using forceps and cut with about 5-6 mm of scleral
155 tissue around it, then washed with 0.9% NaCl solution. Then the cornea was placed between
156 the donor compartment and the Franz diffusion cell receptor. The receptor compartment with
157 a capacity of 10 ml was filled with STF and stirred at 100 rpm stirring on a magnetic stirrer.
158 The temperature was maintained at $37 \pm 0.5^\circ\text{C}$. The formulation was applied into the donor
159 compartment. At 0.25, 0.5, 0.75, 1, 2, 3, 4, 5, 6-, 7-, 8-, and 24-hours, samples (1 mL each)
160 were sampled and replaced with equal volumes of receptor medium. The absorbance was
161 measured at a wavelength of 233.4 nm. Then, the drug concentration in the sample was
162 calculated [18]. Afterwards, the fluxes (J , $\mu\text{g}/\text{cm}^2/\text{h}$) and the permeability coefficients (K_p ,
163 cm/h) were determined as previously described [23,24]

164 **2.7. Hemolysis Study**

165 Fresh blood was collected, anticoagulated, and then stored at room temperature [25].
166 The red blood cells were collected and diluted with PBS pH 7.4 to achieve a final
167 concentration of 10% v/v. Then, the test formulation was incubated at with the red blood

168 cells. The hemolysis percentages of the hemoglobin from the blood were measured as
169 reported previously [20,26].

170

171 **2.8. In vitro antibacterial activity**

172 **2.8.1. Culture of *Pseudomonas aeruginosa***

173 *Pseudomonas aeruginosa* (ATCC® 9027) was purchased from LGC Standards,
174 Middlesex, UK. The bacterial were cultured in tryptic soy broth (TSB) media at 37°C
175 overnight. The cultured bacterial were diluted in fresh TSB to obtain a final bacterial number
176 of 1×10^8 CFU/mL. The optical density was measured at 550 nm.

177 **2.8.2. In vitro antibacterial activity of CFZ in mucoadhesive-thermosensitive *in situ* gel 178 formulation**

179 Agar diffusion method was used to assess the antibacterial activity of CFZ following
180 the incorporation into mucoadhesive-thermosensitive *in situ* gel formulation. The bacterial
181 suspension of *Pseudomonas aeruginosa* was cultured in tryptic soy agar (TSA) using
182 spreading method. Following this, paper discs containing CFZ solution, CFZ thermosensitive
183 formulation, water and free disc were placed on the top of the bacterial culture. The
184 concentration of CFZ used was equal to 5 µg/mL. The media was incubated at at 37°C
185 overnight and the inhibitory zone was calculated using digital vernier calipers.

186

187 **2.9. In vivo irritation study on rabbit's eye**

188 This study was carried out under the permission from Health Ethical Committee of
189 Hasanuddin University, Indonesia. In this study, the application of the formulation was
190 performed in the rabbit's eyes without the use of anesthesia. Several assessments of signs of
191 irritation including redness, swelling, cloudiness, edema, bleeding, and discharge were
192 observed at regular intervals during the 3 days of treatment. Draize eye irritation test was
193 used to observe changes in the cornea, conjunctiva, and iris in the rabbit's eyeball after
194 exposure to test substances. In this study, 0.1 mL of the formulation was applied on the cornea

195 and conjunctival sac of one rabbit's conscious eyeballs. The irritation of the eye was assessed
196 using Maximum Average Score (MAS) [27].

197 **2.10. Histopathological evaluation**

198 In the end of the experiments, the rabbits were dissected for their ocular tissue
199 collection and preserved preserved in 10% neutral buffered formalin. Then, the tissue was
200 placed in paraffin wax. Sections with a thickness of 4 micrometers were cut and stained with
201 Hematoxylin and Eosin (H&E) and then observed using a microscope, as previously
202 described [28].

203 **2.11 Statistical analysis**

204 All data were reported as mean \pm standard deviation (SD) and statistically analyzed using
205 one-way ANOVA and Tukey's test with GraphPad Prism version 6.0 (GraphPad Software,
206 CA, USA).

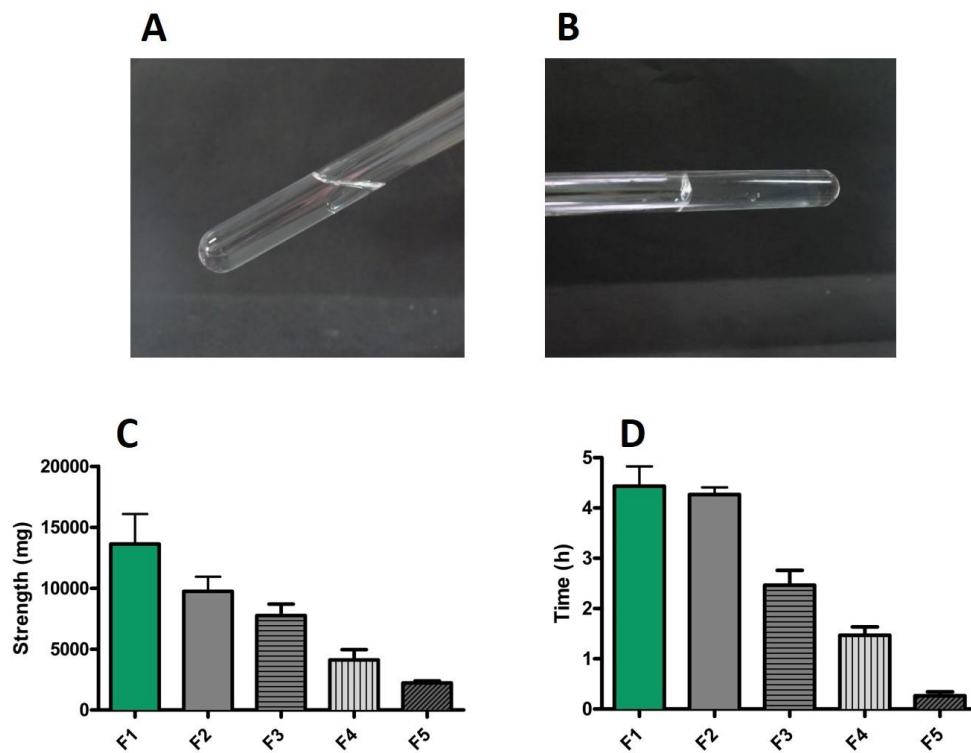
207 **3. Result and Discussion**

208 **3.1. Mucoadhesive thermosensitive *in situ* gel formulation of CFZ**

209 The *in situ* ocular gels were formulated using various concentration of Pluronic[®] F127 and
210 Pluronic[®] F68. Initially, several formulations were prepared to achieve the formulation which
211 could form a liquid in the room temperature and change into a gel in the body temperature.
212 It was found that the different ratio of both polymers resulted in different gelation
213 temperatures. The various of gelation temperature was mainly affected by the concentration
214 of Pluronic[®] F127 and F68 on the gel. The gelation temperature decreased along with the
215 increase of Pluronic[®] F127 and decreased in the amount of Pluronic[®] F68. This phenomena
216 could explained as Pluronic[®] or poloxamer is generally consist of two monomer, poly-
217 ethylene-oxide (PEO) and poly-propylene-oxide (PPO). These two monomers play an
218 important role in affecting gelation temperature, especially PPO block [29]. A decrease of
219 PPO block resulted in an increase of gelation temperature. Pluronic[®] 127 127 has 30% of
220 PPO while Pluronic[®] F68 consist of 20% of PPO. Addition of concentration of Pluronic[®]
221 F88 into the formula showed a decrease of cefazolin sol-to-gel temperature. Following our
222 preliminary studies, the ration of 15:5 of Pluronic[®] F127 and Pluronic[®] F68 could form a gel

223 in the body temperature, while form liquid in the room temperature. Therefore, this ratio was
224 used in the further studies. Total of 50 grams each formula was obtained, with the appearance
225 of being clear and liquid at cold temperatures and then turning into a gel when heated (Figure
226 1A and 1B).

227 It was important to determine the gelation temperature in *in situ* gel formulations in
228 combination with or without HA as a mucoadhesive agent used in the formulation. The
229 acceptable gelation temperatures value close to body temperature were obtained, which were
230 in the range of 30-37°C. The five formulations have gelation temperature values in the range
231 of 31.3°C to 36.7°C. In addition, the five formulations were tested for gelation temperature
232 again by dissolving them using STF to investigate whether the tear fluid could affect the
233 gelation temperature of the five formulations. The five formulations that have been diluted
234 with STF possessed gelation temperature values in the range of 30.7°C to 36.3°C. Based on
235 the evaluations, desired results were obtained with the five formulations having gelation
236 temperature values which were in the optimum gelation temperature range of 30-37°C [10].
237



238

239 **Figure 1.** CFZ mucoadhesive thermosensitive *in situ* gel formulation in room temperature (A) CFZ
 240 mucoadhesive thermosensitive *in situ* gel formulation in body temperature (B). Mucoadhesive strength (C) and
 241 mucoadhesive time (D) of five formulations of mucoadhesive thermosensitive *in situ* gel of CFZ (mean \pm
 242 SD, $n = 3$).
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 244

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245

3.2. Mucoadhesive study of CFZ *in situ* gel formulation

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252

In several years, the mucoadhesive concept has very often become a theory that has attracted a lot of attention, especially on its effectiveness in increasing the bioavailability of drugs [30]. There are many theories that discuss this mucoadhesive phenomenon, one of the theories is the adsorption theory. This theory discusses the mechanism of mucoadhesion, including hydrogen bonds, van der Waals, and hydrophobic agents. Although this type is classified as weak bound, but if many bounds were formed, it could produce a strong level of adhesion [30].

253

254

255

In this study, there were two important aspects that were discussed in the mucoadhesive study, namely mucoadhesive strength and mucoadhesive time. Mucoadhesive strength represents the amount of weight given until the formula was separated from the tissue to

256 observe the level of adhesion of the formulation. On the other hand, mucoadhesive time is
257 the time required by the formulation for separating between the tissue when given some
258 weight. The results obtained from the five formulations are depicted in Figure 1C and 1D.

259 As can be seen in Figure 1C and 1D, F1 showed the largest value in both aspects,
260 strength and time, while F5 showed the lowest value. This was due to the variation in the
261 concentration of mucoadhesive agents, namely HA, where F1 possessed the highest HA
262 concentration while F5 was not mixed with HA. The value of F5 has a mucoadhesive strength
263 value of 13625.36 ± 2474.52 mg and a mucoadhesive time of 4.43 ± 0.39 h. This value was
264 the highest value between five formulations. This could potentially result in the increase of
265 drug released in the eye tissue, leading to the effectiveness of the treatment. From the
266 statistical analysis, it was also found that the *p* value was less than 0.05, indicating that HA
267 concentrations significantly affected the mucoadhesive properties of the formulations.

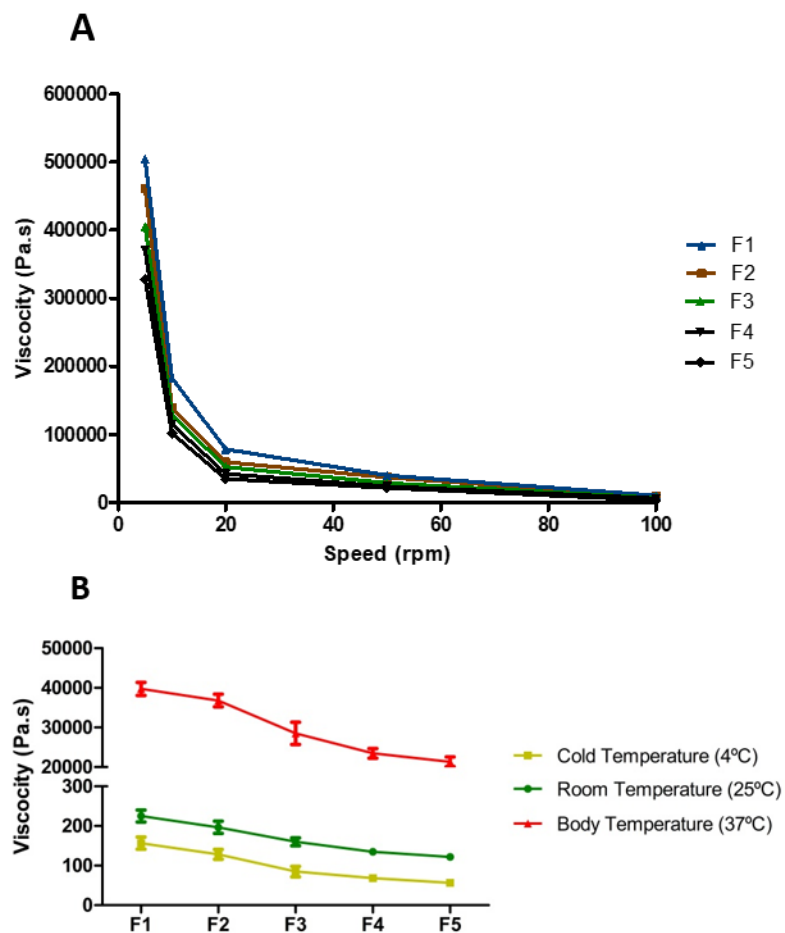
268 3.3. pH Evaluation

269 The five formulations had pH values in the range of 6.69 to 6.83 and were therefore
270 acceptable for ophthalmic applications [31].

271 3.4. Viscosity of formulation at three different temperature and rheology study of gel *in* 272 *situ* formulation

273 Polymers, however, are non-Newtonian fluids, and most often exhibit shear-thinning
274 behavior, which is entangled, the long molecules begin to unravel and are oriented along the
275 flow direction when the applied deformation is high enough. Shear thinning is a property
276 which eventually makes many processing methods possible [32]. In this study, the viscosity
277 of the thermosensitive gels was observed at 4 °C (cold temperature), 25 °C (room
278 temperature), and 37 °C (body temperature) (Figure 2). Ideally, the formulation should
279 possess the property which could form a free-flow liquid below body temperature in order to
280 ease the administration of the formulation while changing into a semisolid preparation at
281 temperature of the body in order to improve the contact time. With respect to the
282 characteristic of the rheology, commonly, the gel preparation shows pseudoplastic flow.
283 Accordingly, the thermosensitive formulation should be able to produce shear-thinning
284 properties at room and body temperatures. In this type of rheological property, the viscosity

285 of the formulation would decrease following the improvement of the shearing stress. This was
286 due to the theory that the high velocity could influence the three-dimensional structures in
287 the gel system, thereby reduce the low viscosity [19,29]. As shown in Figure 2, all
288 formulations possessed this characteristic. Accordingly, the formulations developed in this
289 study could meet the criteria of the desired formulation. It was also found that the addition
290 of HA as mucoadhesive agent did not influence the thermosensitive properties of the
291 formulations.



292

293 **Figure 2.** Rheology of CFZ mucoadhesive-thermosensitive *in situ* gel formulation with variety of spindle
294 rotation (rpm) (A). Viscosity of CFZ mucoadhesive-thermosensitive *in situ* gel formula in three different
295 temperatures (B) (mean \pm SD, $n = 3$).
296

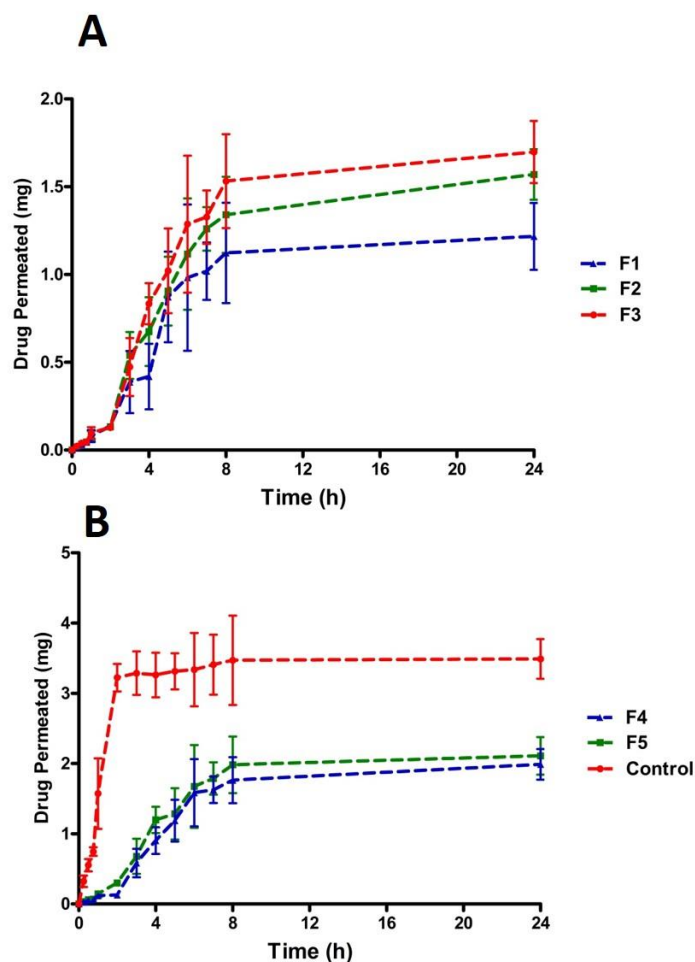
297 **3.5. Drug content determination of CFZ in mucoadhesive-thermosensitive *in situ* gel**
298 **formulation**

299 In an attempt to investigate whether the drug content in the formula was in agreement with
300 the actual content, it was necessary to test the drug content. In this study, the percent of
301 recovery drug was carried out, all formulations were determined for the percent recovery by
302 comparing the levels of the CFZ stock solution of 10 µg/mL. It was found that the percent
303 recovery values of the five formulations ranged from $98.33 \pm 2.47\%$ to $101.43 \pm 2.60\%$. From
304 these results, it can be concluded that the CFZ content in the formula was corresponded with
305 the concentration used in the formula, and the results obtained can be accounted for the
306 further tests.

307 **3.6. *In vitro* drug release study**

308 This step aimed to evaluate the amount of released drug form the thermosensitive
309 formulations. The drug content remaining in the ocular tissue is very important for the
310 effectiveness in the treatment of bacterial keratitis. In this study, the amount of released from
311 mucoadhesive-thermosensitive *in situ* gel formula was compared to the solution form. It was
312 found that in *in vitro* release study (Figure 4), only after 3 hours, the amount of released drug
313 of control solution was 3.17 ± 0.23 mg. On the other hand, the amount of released drug in 24
314 hours for the formulations F1, F2, F3, F4, and F5 were 1.22 ± 0.19 mg, 1.57 ± 0.14 mg, 1.70
315 ± 0.18 mg, 1.99 ± 0.22 mg, and 2.11 ± 0.27 mg, respectively.

316 From the result obtained in this study, it could be hypothesized that CFZ formulated
317 into thermosensitive *in situ* gel was capable of controlling the release of CFZ across the
318 dialysis membrane. As this formulation could control the release of the drug, the longer the
319 drug in the membrane could potentially improve the effectiveness of the treatment for
320 keratitis. It was also found that the addition of HA could reduce the drug release profiles.



321

322 **Figure 3.** *In vitro* permeation study of CFZ thermosensitive mucoadhesive *in situ* gel in dialysis membrane
 323 (mean \pm SD, $n = 3$).

324

325 It was important to investigate the release mechanism of CFZ from mucoadhesive-
 326 thermosensitive *in situ* gel formulation in *in vitro* release study. The release mechanism was
 327 performed by fitting the data result to different mathematical kinetic models with DD solver
 328 app [33]. The release mechanism was determined based on the correlation determination (R^2)
 329 value nearly to ($R^2 = 1.00$), which the correlation determination of *in vitro* study can be seen
 330 on Table 2.

331

332 **Table 2.** *In vitro* correlation coefficient of CFZ from mucoadhesive-thermosensitive in situ gel
 333 formulation in different mathematic kinetic models

Mathematic model	Correlation Determination (R ²)				
	F1	F2	F3	F4	F5
Zero Order Kinetics	0,5930	0,6514	0,6274	0,6187	0,5935
First Order Kinetics	0,4387	0,5148	0,4903	0,4842	0,4100
Higuchi Models	0,7754	0,8189	0,7995	0,7909	0,7970
Krossmeyer – Peppas Model	0,2506	-0,5205	0,2146	0,2096	-0,4888
Hixson – Crowell Model	0,4676	0,3980	0,3699	0,3217	0,3891

334

335 Following the calculation of the mathematic models, all formula followed Higuchi's
 336 model for the release mechanism with correlation coefficient of F1, F2, F3, F4 and F4
 337 respectively were 0,7754; 0,8189; 0,7995; 0,7909; 0,7970. The Higuchi's model indicates
 338 that the drug release in this system was gradually influenced by the swelling the matrix [34].
 339

340 **3.7. *Ex vivo* permeation study of CFZ mucoadhesive-thermosensitive in situ gel** 341 **formulation**

342 It is important to consider that the amount of drug concentration in the ocular tissue is very
 343 influential for the effectiveness of treatment for bacterial keratitis therapy. In this case, the
 344 drug was permeated through the *ex vivo* cornea tissue and the results are depicted in Figure
 345 4A. From the data, it was shown that after 24 hours, control solution could produce $3.85 \pm$
 346 0.37 mg of permeation amount. Furthermore, in the case of thermosensitive *in situ* gel, the
 347 amount of CFZ permeating after 24 hours was significantly lower ($p < 0.05$) compared to
 348 control solution which was in the range between 1.01 ± 0.11 mg to 1.80 ± 0.21 mg. In
 349 agreement with *in vitro* results, the mathematic model of the *ex vivo* permeation also followed
 350 Higuchi's model with determination coefficient was 0,7754; 0,8189; 0,7995; 0,7909; and
 351 0,7970 respectively (Table 3). Moreover, flux of permeation (J_{ss}) and permeation coefficient
 352 (K_p) of CFZ from mucoadhesive-thermosensitive were calculated and compared to the

353 control solution (Table 4). It was found that the J_{ss} and K_p values of control solution was
 354 significantly higher ($p < 0.05$) compared to thermosensitive formulations. Accordingly,
 355 considering the values and kinetic model results, similar to *in vitro* drug release study, this
 356 study depicted that the incorporation of CFZ into thermogelling system could control the
 357 release the permeation through the ocular tissue.

358

359 **Table 3.** *Ex vivo* determination coefficient of CFZ from mucoadhesive-thermosensitive

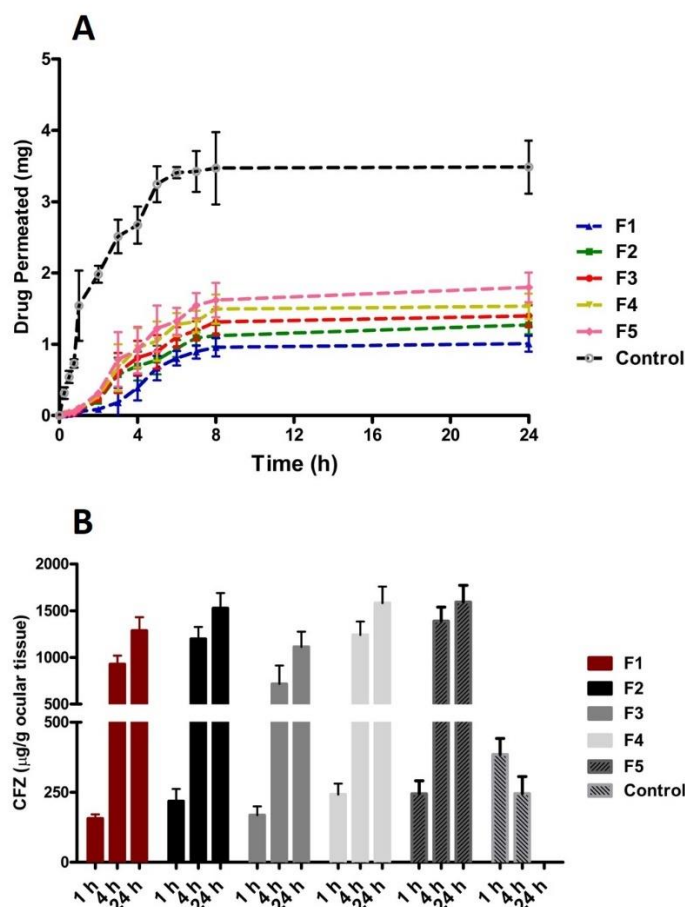
Mathematic model	Correlation Coefficient (R^2)				
	F1	F2	F3	F4	F5
Zero Order Kinetics	0,5930	0,6514	0,6274	0,6187	0,5935
First Order Kinetics	0,4387	0,5148	0,4903	0,4842	0,4100
Higuchi Models	0,7754	0,8189	0,7995	0,7909	0,7970
Krossmeyer – Peppas Model	0,2506	-0,5205	0,2146	0,2096	-0,4888
Hixson – Crowell Model	0,4676	0,3980	0,3699	0,3217	0,3891

360

361 For the further analysis, the *ex vivo* retention was measured. The *ex vivo* retention was
 362 carried out to observe the amount of drug content localized in the corneal tissue which could
 363 be used to estimate the effectiveness of this approach [11]. This was performed by calculating
 364 the concentration level retained in the tissue after 1, 8, and 24 hours of the application of each
 365 formulation. The retention results were also compared to control solution and the results are
 366 exhibited in Figure 4B.

367

368



369

370 **Figure 4.** *Ex vivo* permeation study of CFZ mucoadhesive thermosensitive *in situ* gel in ocular tissue compared
 371 with control eye drop (A). *Ex vivo* retention of mucoadhesive thermosensitive *in situ* gel and control eye drop
 372 of CFZ in ocular tissue (B) (mean \pm SD, $n = 3$).

373

374 According the bar charts, it was shown that the incorporation CFZ to the
 375 thermosensitive hydrogel could increase the location of CFZ in the ocular tissue. The amount
 376 of CFZ localized in the ocular tissue increased from 1 h to 24 h after the application of the
 377 formulation. After 24 h, the amount of CFZ detected was in the range of 1287.11 ± 143.72
 378 $\mu\text{g/g} - 1592.19 \pm 179.34 \mu\text{g/g}$. Interestingly, no CFZ was detected in the case of control
 379 solution after 24 h of application. Therefore, the thermosensitive system could not only
 380 control the release of CFZ, but also improve the concentration of CFZ in the ocular tissue,
 381 which could potentially improve its effectiveness in the treatment of bacterial keratitis.

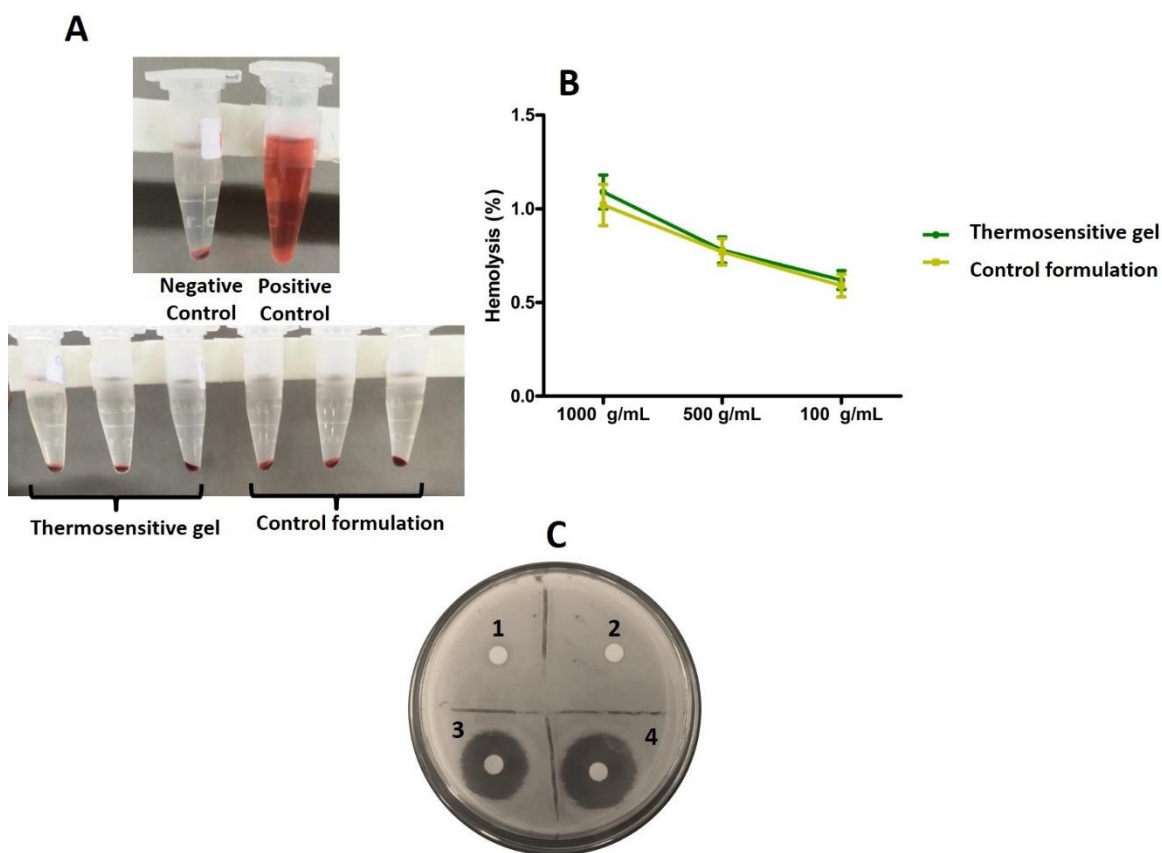
382 **Table 4.** Flux of permeation (Jss) and permeation coefficient (Kp) of CFZ from mucoadhesive-thermosensitive
383 and control solution (mean \pm SD, $n = 3$).

Formulation	Jss ($\mu\text{g}/(\text{cm}^2\text{h})$)	Kp (cm/h)
F1	127.23 \pm 11.98	0.054 \pm 0.001
F2	133.93 \pm 10.39	0.065 \pm 0.001
F3	149.11 \pm 12.13	0.078 \pm 0.002
F4	154.17 \pm 13.09	0.087 \pm 0.002
F5	176.27 \pm 16.12	0.093 \pm 0.001
Control solution	687.23 \pm 41.23	0.233 \pm 0.01

384

385 **3.8. Hemolysis study**

386 It was crucial to investigate the potential toxicity of the new formulation. In this study, we
387 used hemolysis parameter of the blood red cells as initial evaluation to investigate the
388 potential toxicity which could be caused by the formulations. Several studies have used this
389 evaluation to screen the potential of toxicity of numerous drug delivery systems
390 [20,26,35,36]. The results of this study are exhibited in Figure 5A and 5B, showing that the
391 hemolysis percentages were below 1.5% in all concentrations tested. Accordingly, the
392 formulations developed in this study were considerably safe.



393

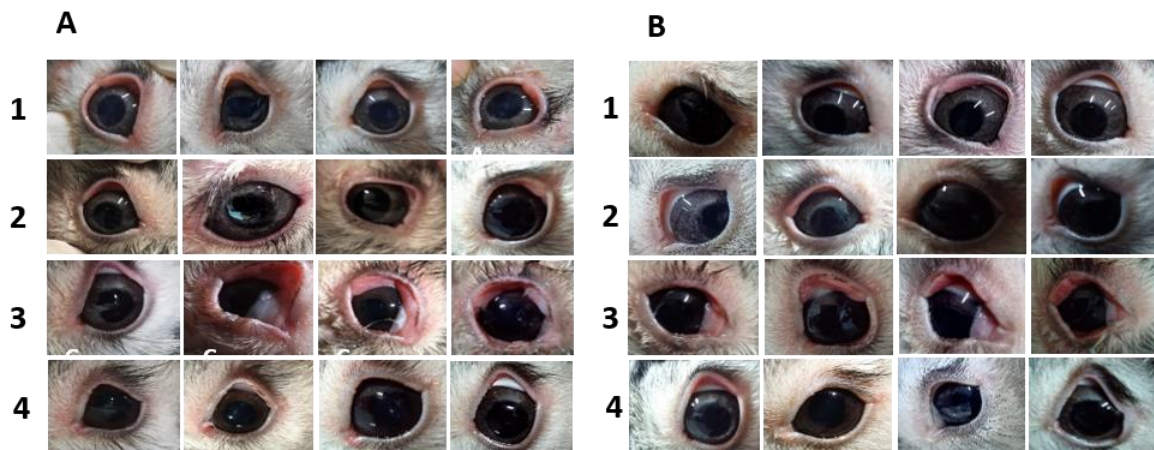
394 **Figure 5.** Representative images of *in vitro* hemolysis study in red blood cells (A). Hemolysis percentages of
 395 thermosensitive and control formulations (B) (mean \pm SD, $n = 3$). Inhibition zones of blank disc (1), water (2),
 396 CFZ solution (3) and CFZ thermosensitive formulation against *Pseudomonas aeruginosa* (C).

397

398 **3.9. *In vitro* antibacterial activity**

399 It was critical to ensure that the incorporation of CFZ into thermosensitive *in situ* gel did not
 400 reduce its antimicrobial activity. In this study, we evaluated the antimicrobial activity using
 401 agar diffusion method against *Pseudomonas aeruginosa*. Figure 5C depicts the representative
 402 image of this study. It was found that the inhibition zones of CFZ solution and CFZ in
 403 thermosensitive *in situ* gel were 21.98 ± 1.98 mm and 20.76 ± 1.21 mm. Analyzed
 404 statistically, there was no a significant different ($p > 0.05$) between those values, indicating
 405 that the formulation did not affect the antimicrobial activity of CFZ against *Pseudomonas*
 406 *aeruginosa*.

407 **3.10. In vivo irritation evaluation**



408

409 **Figure 6.** Rabbit's eye on day one treatment (A) and last day treatment (B), (1 = with CFZ), (B = With NaCl
410 0,9%), (C = with Sodium lauryl sulfate (SLS)), (D = without treatment).

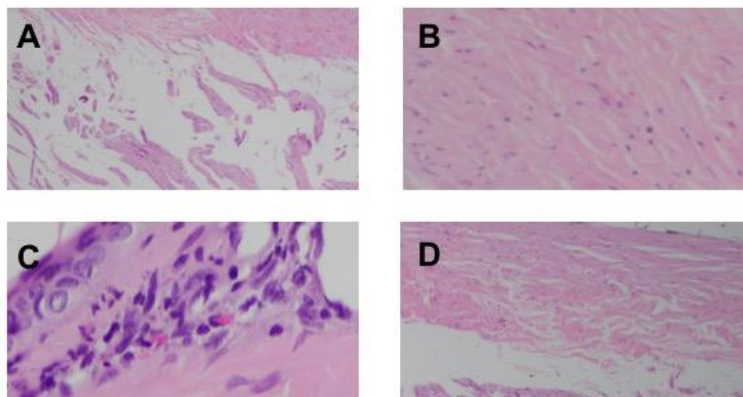
411

412 We further assess the irritation of our approach *in vivo* in rabbits. Figure 6 shows the result
413 of the treatment after 4 days. The rabbit eye with CFZ in situ gel showed a comparable result
414 compared to the rabbit eye without treatment and the administration of NaCl 0,9%. On the
415 other hand, following the administration of SLS, the rabbit's eyes were observed to be red
416 and the ciliary muscles were swollen. Therefore, the approach developed in this study were
417 not found to be irritable to the ocular tissue following the multiple day application.

418

419 **3.11. Histopathological evaluation**

420 Following the irritation evaluation, we collected the ocular tissues and observed using
421 histopathological examinations. The results of this study are depicted in Figure 7. It was
422 found that in the administration of the formulation, NaCl 0.9% and no treatment groups, no
423 infiltration, congestion and edema were observed. In contrast, infiltration, congestion and
424 edema were all observed after the administration of SLS. Therefore, the thermosensitive
425 formulations were potentially safe for ocular administration.



426

427 **Figure 7.** Histopathological result of (A) NaCl 0,9% solution; (B) CFZ mucoadhesive-thermosensitive in situ
428 gel formula; (C) sodium lauryl sulfate (SLS) solution; and (D) without treatment

429

430 Overall outcomes have proven that the incorporation of CFZ into themosensitive-
431 mucoadhesive hydrogel could potentially improve the localization of CFZ in the ocular
432 tissue, while controlling the release behavior. This approach could potentially be beneficial
433 in the treatment of bacterial keratitis as it was also found that no irritations were observed
434 following the multiple day applications, and importantly, no potential toxicity was found.

435 The use thermosensitive hydrogel in ocular drug delivery have been widely explored.
436 However, several aspects should be considered. As this was intended to ocular
437 administration, before moving to the industrial steps, sterilization process should be
438 developed. Several studies have shown that Pluronic[®]-based hydrogels could be sterilized
439 using steam heat sterilization with autoclave [37,38]. However, due to the instability of CFZ
440 in high temperature, gamma radiation could be the appropriate method for the sterilization.

441 The development of this system is still in the early step. Therefore, it is also crucial to further
442 optimize the formulation to ensure that the hydrogel possess adequate properties after the
443 administration to the patients. As the formulation is in the liquid form, the hydrogel could be
444 administered using the same method as the eye drops. Additionally, the information
445 regarding the transformation of the liquid to the gel after the administration should be given
446 to the patients. However, prior to the application in the clinic, further study investigating *in*

447 *vivo* pharmacokinetic and pharmacodynamic studies in appropriate animal models should
448 now be conducted.

449

450 **4. Conclusion**

451 Based on all evaluation results of the mucoadhesive-thermosensitive *in situ* gel of CFZ,
452 it can be concluded that the CFZ *in situ* gel preparation could increase the contact time of the
453 preparation with the eye up to 4 hours. The formulations were able to sustain the release of
454 CFZ and improve the retention in the eye tissue when compared to control solution with no
455 significant change in antimicrobial activity. Following the hemolysis assay, the formulation
456 was potentially safe. Importantly, the *in situ* gel preparation also did not irritate the
457 experimental animals applied with the *in situ* gel preparation.

458

459 **Author contributions:**

460 **Muh. Al Fiqri:** Conceptualization, Methodology, Funding acquisition, Writing – original
461 draft. **Alhidayah:** Methodology, Writing – original draft. **Nirmayanti:** Methodology,
462 Writing – original draft. **Ummu Athiyah:** Methodology, Data curation. **Patricia**
463 **Layadi:** Methodology, Data curation. **Tamara Gabriela Angeleve Fadjar:** Data curation,
464 Validation **Andi Dian Permana:** Conceptualization, Project administration, Funding
465 acquisition, Validation, Supervision.

466 **Declaration of Competing Interest**

467 The authors declare that they have no known competing financial interests or personal
468 relationships that could have appeared to influence the work reported in this paper.

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